

Research Article

Degradation Study of Ciprofloxacin Hydrochloride, Bromhexine hydrochloride and their Combined Pharmaceutical Dosage Form by Spectrophotometry

Avinash M Nagapara^{1*}, Hasmukh C Nagapara², Darshan Madiya¹, Shital Faldu¹ ¹Department of Quality Assurance, Smt. R. D. Gardi B. Pharmacy College, Rajkot, Gujarat ²Virani Science College, Rajkot, Gujarat, India *avinashnagapara@yahoo.com



ABSTRACT

A stress Degradation study of the combination containing Ciprofloxacin Hydrochloride and Bromhexine Hydrochloride was carried out by various reagent like 1 M NaOH, 1M HCl, 6% H_2O_2 and Neutral water at higher temperature. This combination is widely used for the treatment of various types of respiratory disorders and COPD. The effects of the various stress conditions were observed in terms of decrease in the peak height, increase in peaks or slightly change or shifting of the wavelengths. It was found that by applying various stress conditions, 4% to 50% drugs were degraded in case of pure drugs as well as the Commercial formulation. Thus spectrophotometry was successfully utilised for primary stability study of the pure drug as well as the degradation study of Ciprofloxacin Hydrochloride and Bromhexine Hydrochloride.

Keywords: Degradation study, Bromhexin hydrochloride, Respiratory disorders, Ciprofloxacin hydrochloride, Spectrophotometry

INTRODUCTION

Ciprofloxacin HCl (fig. 1a) is the salt of Ciprofloxacin HCl, chemically known as, 1cyc1opropyl-6-fluoro-1, 4- dihydro-4-oxo-7(1 piperazinyl)-3-quinolinecarboxylic acid hydrochloride monohydrate^[1]. it is a first generation fluoroquinolone. Its spectrum of activity includes most strains of bacterial pathogens responsible for respiratory, urinary abdominal tract, gastrointestinal, and infections, including Gram-negative and grampositive bacterial pathogens^[2]. Bromhexine is a synthetic derivative of the herbal active ingredient vasicin, they chemically known as 2amino-3,5-dibromobenzylm(cyclohexyl)methyl amine hydrochloride^[1]. CIPRO is official in USP, BP, EP & IP whereas BROM is official in USP, BP, EP. The chemical structures of CIPRO & BROM are shown in Fig. 1.Combination drug products of CIPRO and BROM are widely marketed and used in the treatment of mucous plugs, patient with chronic obstructive lung disease and bronchitis.

Some procedures have been reported for the degradation study of CIPRO and BROM in single dosage forms. But for the combination no one method is found. From the literature, no degradation studies could be founded for BROM and CIPRO combination. To check the degradation of the drugs by applying various stress conditions as well as the stability study of various drugs is one of important criteria for an analyst ^[2-4]. Therefore degradation study of this combination in combined dosage forms seemed to be necessary. An experiment was made to find the degradation of these drugs individually

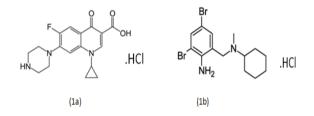
How to cite this article: AM Nagapara, HC Nagapara, D Madiya, S Faldu; Degradation Study of Ciprofloxacin Hydrochloride, Bromhexine hydrochloride and their Combined Pharmaceutical Dosage Form by Spectrophotometry; PharmaTutor; 2014; 2(7); 102-109

PharmaTutor Magazine | Vol. 2, Issue 7 | magazine.pharmatutor.org



ISSN: 2347-7881

and with their combination using various stress conditions like refluxing the drugs with high temperature using highly acidic, alkaline or neutral conditions.



MATERIALS AND METHODS

Site of experimentation: Smt. R. D. Gardi B. Pharmacy College, Nyara, Rajkot

INSTRUMENTATION: UV visible spectrophotometer Helios Alpha, Thermo Scientific, (model UV A 1002E) with 1 cm matched quartz cells were used for all absorbance measurements. Contech, EIE instrument pvt. Ltd. (model CA 34) balance was used for weighing the samples

REAGENTS: Double distilled water and Whatmann filter paper (0.45µm) were used for filtration. Active pharmaceutical ingredient (API) of Ciprofloxacin HCl (CIPRO), Bromhexine HCl (BROM) were obtained as gift sample from Aarti drugs,Mumbai and Ven petrochem & pharma PVT. LTD. (tablets with composition BROM-8 mg and CIPRO-250 mg) were procured from the local market.

CHEMICAL: Methanol, NaOH, Hydrogen Peroxide and HCl are obtained from local market.

SOLUTION:

Stock solutions, 1 mg mL-1 of pure samples of CIPRO and BH were freshly prepared individually in methanol. For acidic degradation, 1M HCl solutions were prepared. To study neutral degradation, doubled distilled water was used for preparing various solutions. To study alkaline degradation, 1M NaOH solutions were prepared. For oxidative degradation, 6% H2O2 were used.

PROCEDURE:

1ml of stock solution of pure CIPRO, (1 mg mL-1), was taken to a round bottom flask containing 9ml of 1M HCl. After ten times dilution, the absorbance of resultant solution was taken using blank solution that contained all except the drug and was named as zero time reading. Then solution was reflux at 60°C for 120 minutes on an oil bath.

During refluxing process, the samples were taken after 30 minutes, 60 minutes and 120 minutes intervals. Then the resultant solutions were cooled and absorbance of the resultant solutions were measured using blank treated by the same way after ten time dilutions of each. The same procedure was applied for different degradation conditions like refluxing with neutral condition using distilled water, 1M NaOH, 6% H₂O₂. The overall procedure was followed for pure BH, pure CIPRO and commercial formulation of both the drugs. The solution of commercial formulation was prepared in 1: 1 ratio by standard addition method. The table 1 shows the sampling plan total degradation study.

Type of Degradation	Solution used for degradation study	Sampling plan
Acidic degradation	Reflux at 60°C with 1M HCl	Initial, 60 minutes, 120 minutes
Neutral degradation	Distilled water, reflux at 60°C	Initial, 60 minutes, 120 minutes
Alkaline degradation	Reflux at 60°C with 1M NaOH	Initial, 60 minutes, 120 minutes
Oxidative degradation	6% H2O2 and reflux at 80°C	Initial, 60 minutes, 120 minutes

Tables 1 Sampling plan for estimation of degraded products ^[5-9].



RESULT AND DISCUSSION Table 2 The results of the acidic degradation studies were observed in the table given below: Condition Drug Amt. of drug degraded **Ciprofloxacin HCl** After refluxing with 25.98 % ciprofloxacin 1M HCl at 60°C for- Initial 60 hydrochloride is degraded after 120 minute refluxing. minute and 120 minute **Bromhexine HCl** After refluxing with 31.73 % bromhexine 1M HCl at 60°C for- Initial 60 hydrochloride is degraded minute and 120 minute after 120 minute refluxing. Combination (Cinor-BR) After refluxing with In combination 12.05 % and 1M HCl at 60°C for-13.35 % degradation of Initial 60 minute and ciprofloxacin hydrochloride 120 minute and bromhexine hydrochloride respectively.

By applying highly acidic conditions,

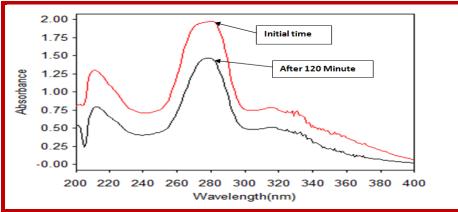


Figure 1 degradation of CIPRO in acidic condition.

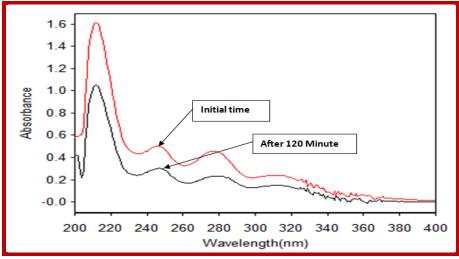


Figure 2 degradation of BROM in acidic condition.



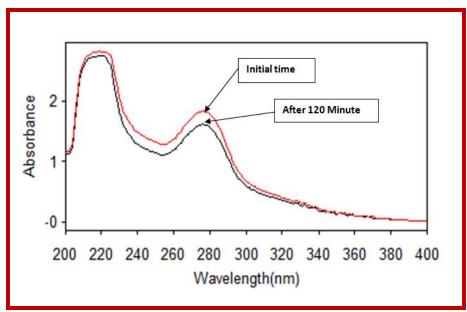


Figure 3 Degradation of CIPRO and BROM in acidic condition.

Table 3 The results of the neutral degradation studies were observed in the table given below:
--

Drug	Condition	Amt. of drug degraded
Ciprofloxacin HCl	After refluxing with Distilled	4.31 % ciprofloxacin
	water at 60°C for- Initial 60	hydrochloride is degraded
	minute and 120 minute	after 120 minute refluxing.
Bromhexine HCl	After refluxing with Distilled	8.75 % bromhexine
	water at 60°C for- Initial 60	hydrochloride is degraded
	minute and 120 minute	after 120 minute refluxing.
Combination (Cinor-BR)	After refluxing with Distilled	In combination 13.57 % and
	water at 60°C for- Initial 60	4.31 % Degradation of
	minute and 120 minute	ciprofloxacin hydrochloride
		and bromhexine
		hydrochloride respectively.

By applying Neutral conditions,

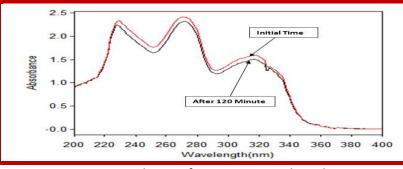


Figure 4 Degradation of CIPRO in neutral condition.



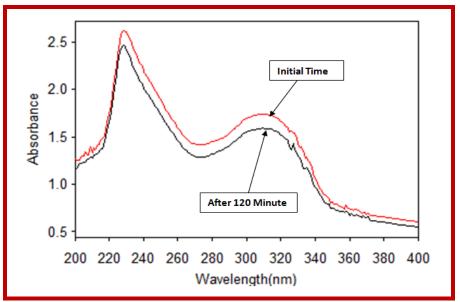


Figure 5 Degradation of BROM in neutral condition.

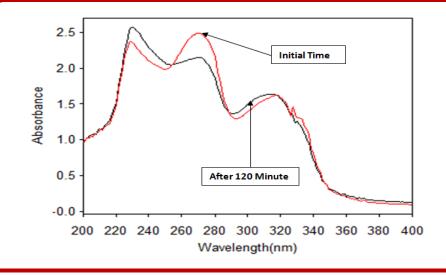


Figure 6 Degradation of CIPRO and BROM in neutral condition.

Drug	Condition	Amt. of drug degraded
Ciprofloxacin HCl	After refluxing with 1M	47.3 % ciprofloxacin
	NaOH at 60°C for- Initial 60	hydrochloride is degraded
	minute and 120 minute	after 120 minute refluxing.
Bromhexine HCl	After refluxing with 1M	52.56 % bromhexine
	NaOH at 60°C for- Initial 60	hydrochloride is degraded
	minute and 120 minute	after 120 minute refluxing.



ISSN: 2347-7881			
Combination (Cinor-BR)	After refluxing with 1M	In combination 45.78 % and	
	NaOH at 60°C for- Initial 60	52.91 % degradation of	
	minute and 120 minute	ciprofloxacin hydrochloride	
		and bromhexine	
		hydrochloride respectively.	

By applying Basic conditions,

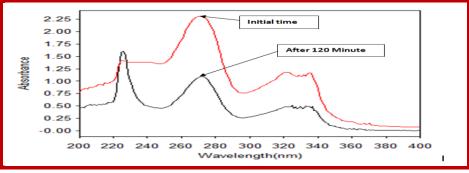


Figure 7 Degradation of CIPRO in basic condition.

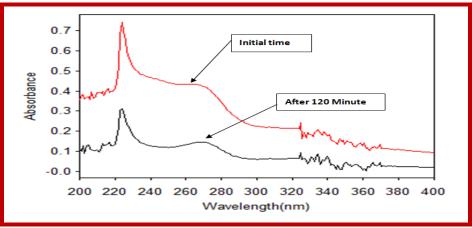


Figure 8 Degradation of BROM in basic condition.

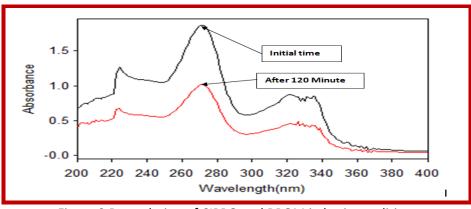


Figure 9 Degradation of CIPRO and BROM in basic condition.



ISSN: 2347-7881

Table 5 The results of the oxidative degradation studies were observed in the table given below:

Drug	Condition	Amt. of drug degraded
Ciprofloxacin HCl	After refluxing with $6 \% H_2O_2$	8.88 % ciprofloxacin
	at 60°C for- Initial 60 minute	hydrochloride is degraded
	and 120 minute	after 120 minute refluxing.
Bromhexine HCl	After refluxing with $6 \% H_2O_2$	4.58 % bromhexine
	at 60°C for- Initial 60 minute	hydrochloride is degraded
	and 120 minute	after 120 minute refluxing.
Combination (Cinor-BR)	After refluxing with $6 \% H_2O_2$	In combination 29.64 % and
	at 60°C for- Initial 60 minute	32.48 % degradation of
	and 120 minute	ciprofloxacin hydrochloride
		and bromhexine
		hydrochloride respectively.

By applying Oxydative conditions,

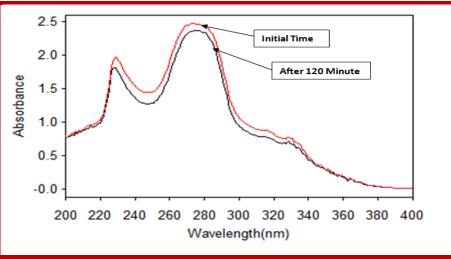


Figure 10 Degradation of CIPRO in oxidative condition.

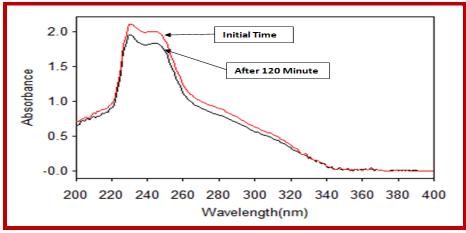


Figure 11 Degradation of BROM in oxidative condition.

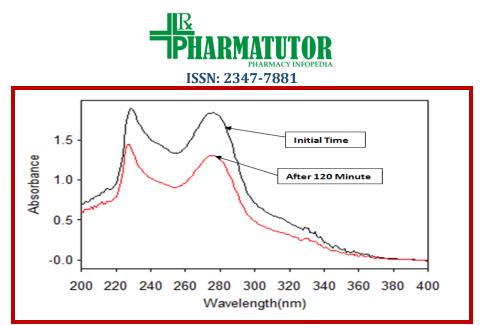


Figure 12 Degradation of CIPRO and BROM in oxidative condition.

CONCLUSION

From the above experiment it was found that degradation was observed in CIPRO , BROM and Combined dosage form. In the study we saw the degradation in the form of decrease in peak height or increase in peak height or change in peak shape. Thus using spectrophotometry, we can get a better result of the degradation of the above combination.

↓ REFERENCES

1. Indian Phamacopoeia, Volume-I & Volume-II, Government of India, Ministry of Health and Family Welfare, Published by The Indian Pharmacopoeia Commission, Ghaziabad, 2007, pp 1092,1093.

2. Rang HP, Dale MM, Ritter JM, Pharmacology, 4th Edn, Churchill Livingston, New York, 1999, pp 282, 344.

3. The United States Pharmacopoeia-30, NF-25, Asian edition, Rockville, MD: The US Pharmacopoeial Convention, Inc., 2004, pp 1756.

4. Tripathi KD, Essentials of Medical Pharmacology; 5thEdn; Jaypee brothers medical publishers, New Delhi, 2004, pp , 619,644.

5. Laine LM, Kaarina KPP and Tammilehto S. "Decomposition of salbutamol in aqueous solutions. The effect of buffer species, pH, buffer concentration and antioxidants." International Journal of Pharmaceutics, 1995, 18, 189-195.

6. Nour TA, Yousry M and Howayda MA. "Potentiometric Flow Injection Analysis of Bromhexine Hydrochloride and its Pharmaceutical Preparation Using Conventional and Coated Wire Ion- Selective Electrodes." Scientia Pharmaceutical, 2006, 74, 121-135.

7. Zaid AN. "Formulation and stability evaluation of 1% w/v oral solution of Bromhexine hydrochloride for veterinary use." The Islamic University Journal, 2007, 15, 13 -22.

8. Kasawar GB, Farooqui M, "Development and validation of a stability indicating RPHPLC method for the simultaneous determination of related substances of albuterol sulfate and ipratropium bromide in nasal solution." Journal of Pharmaceutical and Biomedical Analysis, 2010, 52, 19-29.

9. Hiral ND, Rajeshree CM, "Degradation study of salbutamol sulphate, bromhexine hydrochloride and etofylline in pure and in their ternary mixture by spectrophotometry." International journal of pharmaceutical and health science. 2010, 1, 136-144.